Reference Data

Assignment of ¹H and ¹³C NMR Signals of Eight Apocampholenic Derivatives

M. Findeisen,¹* C. Prehn,² L. Hennig² and K. Schulze²

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ABSTRACT: The ¹H and ¹³C chemical shifts of four apocampholenic aldehydes and the corresponding alcohols are reported. © 1998 John Wiley & Sons, Ltd.

KEYWORDS: NMR; ¹H NMR; ¹³C NMR; sandalwood; apocampholenic derivatives

INTRODUCTION

In connection with investigations on sandalwood fragrances concerning structure-odour relationships, we have recently synthesized the apocampholenic compounds 1–8. One of the analytical aims was the complete assignment of ¹H and ¹³C chemical shifts.

1:
$$R = CH_3$$
 2: $R = C_2H_5$ 5: $R = CH_3$ 6: $R = C_2H_5$

RESULTS AND DISCUSSION

4: $R = C_2H_5$

3: $R = CH_3$

We recorded standard ¹H, ¹³C, ¹³C APT (SEFT), ¹H, ¹H-COSY and ¹³C, ¹H-COSY NMR spectra of all the compounds.

7: $R = CH_3$

8: $R = C_2H_5$

The 1H spectra show signal splittings of higher order in most cases. We determined the center of the multiplets for the chemical shift values if the multiplets were not overlapped by other signals. Otherwise, the use of cross peaks for chemical shift determination made necessary by the complex nature of the spectra leads to a slightly reduced accuracy which we estimate to be about ± 0.01 ppm in the worst cases.

The chemical shifts are compiled for ¹H and ¹³C in Tables 1 and 2, respectively. Scheme 1 shows the numbering of the atoms used in the tables (the numbering was chosen pragmatically and is not in accordance with systematic nomenclature).

It is remarkable that the resonance of C-2 of all compounds is shifted about 10 ppm towards lower field compared with a prediction from a database search (SPECINFO.¹) The latter reports 36.2 ppm with a

standard deviation of 5.5 ppm (the same value for 1–8 owing to the use of only four spheres of neighboring atoms) compared with the experimental values of 46.0 ± 0.2 ppm. Obviously the lack of NMR data for the exact or a similar molecular substructure in the SPECINFO database is the reason for the misleading prediction. The substructures in

Table 1. Chemical shifts of all protons in 1–8 in ppm relative to the solvent $(CDCl_3 = 7.26 \text{ ppm}, \text{ corresponding to TMS} = 0 \text{ ppm})^a$

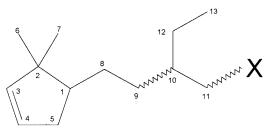
Н	1	2	3	4	5	6	7	8
1	1.69	1.68	1.78	1.77	1.71	1.72	1.97	1.92
3	5.52	5.54	5.56	5.53	5.54	5.55	5.56	5.55
4	5.56	5.56	5.57	5.55	5.56	5.58	5.57	5.55
5	2.42	2.46	2.41	2.42	2.45	2.46	2.44	2.47
5′	1.92	1.93	1.98	1.96	1.95	1.96	2.02	2.01
6	1.05	1.05	1.07	1.06	1.05	1.06	1.11	1.09
7	0.79	0.80	0.85	0.84	0.80	0.81	0.90	0.88
8	1.53	1.47	2.17	2.20	1.79	1.45	2.49	2.45
8′	1.18	1.25	2.00	2.01	1.39	1.22	2.33	2.31
9	1.49	1.42	5.44	5.42	1.33	1.42	6.56	6.45
10	1.63	1.39	_	_	2.36	2.23	_	_
11	3.51	3.57	4.00	4.03	9.64	9.61	9.41	9.35
11'	3.43	3.57	4.00	4.04	_	_	_	_
12	_	1.29	_	2.13	_	1.70	_	2.26
12'	_	1.29	_	2.13	_	1.61	_	2.26
13	0.94	0.91	1.68	1.00	1.11	0.95	1.77	0.96

^a For numbering, see Scheme 1.

Table 2. Chemical shifts of all carbons in 1–8 in ppm relative to the solvent $(CDCl_3 = 77.04 \text{ ppm}, \text{ corresponding to TMS} = 0 \text{ ppm})^a$

\mathbf{C}	1	2	3	4	5	6	7	8
1	49.6	49.5	49.1	49.4	49.3	49.2	48.3	48.6
2	45.8	45.9	45.8	45.9	45.9	45.9	45.9	46.1
3	142.9	142.9	142.6	142.7	142.9	142.9	142.3	142.4
4	127.0	127.0	127.0	127.0	126.9	126.8	126.6	126.8
5	38.0	38.0	37.8	37.8	37.9	37.8	37.6	37.7
6	28.1	28.2	28.1	28.1	28.1	28.1	27.9	28.0
7	21.8	21.9	21.8	21.8	21.8	21.8	21.8	21.9
8	27.2	27.0	27.9	27.6	27.2	27.2	29.5	29.3
9	32.6	29.7	125.9	126.0	30.1	27.9	154.4	154.4
10	36.3	42.4	134.6	140.7	46.7	53.8	139.4	145.4
11	68.2	65.2	68.8	66.9	205.3	205.6	195.1	195.1
12	_	23.6	_	21.1	_	22.0	_	17.3
13	16.8	11.2	13.7	13.1	13.6	11.5	9.1	13.2

^a For numbering, see Scheme 1.



Scheme 1. Generalized structural formula denoting numbering of C atoms used in Tables 1 and 2.

¹ Universität Leipzig, Fakultät für Chemie und Mineralogie, Institut für Analytische Chemie, Linnéstr. 3, D-04103 Leipzig, Germany

² Universität Leipzig, Institut für Organische Chemie, Linnéstr, 3, D-04103 Leipzig, Germany

^{*} Correspondence to: M. Findeisen, Universität Leipzig, Fakultät für Chemie und Mineralogie, Institut für Analytische Chemie, Linnéstr. 3, D-04103 Leipzig, Germany

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the database which are neighbored (neighbored HOSE codes) give 44.8 ± 0.1 ppm (nine lines) and 34.0 ± 0.2 ppm (two lines), leading to the unexpected mean value of 36.2. ppm mentioned above (commented upon by SPECINFO as 'data originate from interpolation between preceding and subsequent code').

An ¹H NOE difference spectrum for 1 indicates that the methyl group at C-2 at lower field (denoted C-6) is in a *trans* position and that at higher field (C-7) is in a *cis* position with respect to the side-chain on C-1

No further stereochemical considerations were made. Each product consists mainly of one stereoisomer (80–90%); exceptions are 1 and 5, where the main portion amounts only to 60%.

EXPERIMENTAL

Compounds

The compounds studied were 2-methyl-4-(2,2-dimethylcyclopent-3-en-1-yl)butan-1-ol (1), 2-ethyl-4-(2,2-dimethylcyclopent-3-en-1-yl)butan-1-ol (2), 2-methyl-4-(2,2-dimethylcyclopent-3-en-1-yl)but-2-enol (3), 2-ethyl-4-(2,2-dimethylcyclopent-3-en-1-yl)but-2-enol (4), 2-methyl-4-(2,2-dimethylcyclopent-3-en-1-yl)butanal (5), 2-ethyl-4-(2,2-dimethylcyclopent-3-en-1-yl)butanal (6), 2-methyl-4-(2,2-dimethylcyclopent-3-en-1-yl)but-2-enal (7), 2-ethyl-4-(2,2-dimethylcyclopent-3-en-1-yl)but-2-enal (8).

Syntheses

In order to produce apocampholenic compounds, we synthesized apocampholenic aldehyde from myrtenal via apopinene and apopinene epoxide.^{2–7} Aldol condensation with propanal and butanal gave the aldehydes 7 and 8 and reduction with LiAlH₄ afforded the allylic alcohols 3 and 4.8.9

The aldehydes $\bf 5$ and $\bf 6$ were obtained by reduction of $\bf 7$ and $\bf 8$ with sodium in liquid ammonia $\bf 10$ and the alcohols $\bf 1$ and $\bf 2$ by further reduction with LiAlH $\bf 4$.

NMR spectra

All spectra were recorded on a Unity-400 spectrometer from Varian operating at 400 MHz (¹H) and 100 MHz (¹³C) at 26 °C and were processed using a SPARC IPX workstation running VNMR software (version 4.1) from Varian.

The ¹H spectra were taken with a 5 mm broadband probe designed for indirect detection and the ¹³C spectra with a normal 5 mm broadband probe. The concentrations were about 5 mg for ¹H NMR and 50 mg for ¹³C NMR in 0.7 ml of CDCl₃ (deuterium lock). The solvent

served as spectrum reference (CDCl $_3=7.26$ ppm for 1H and 77.04 ppm for 13 C). All shifts listed in the tables are rounded to 0.01 ppm for 1H and 0.1 ppm for 13 C.

Typical recording data for ¹H were flip angle 60° (4.5µs), acquisition time 3.8 s, spectral width 5600 Hz (14 ppm), number of scans 32, 41 900 data points for FID and 64K data points for Fourier transformation (digital resolution 0.17 Hz per point), Lorentzian broadening 0.2 Hz. Typical recording data for ¹³C were flip angle 45° (4.5 µs), acquisition time 1.2 s, relaxation delay 3 s, spectral width 25 000 Hz (250 ppm), number of scans 400–1000, WALTZ-16 decoupling of ¹H, 60 000 data points for FID and 64K data points for Fourier transformation (digital resolution 4.24 Hz per point), Lorentzian broadening 2 Hz.

Typical data for APT (attached proton test, SEFT): were similar to those for decoupled $^{13}\mathrm{C}$ spectra with echo delay 7.5 ms.

Typical data for H,H-COSY (90° COSY or 45° COSY in magnitude mode) were: acquisition time 0.247 s (1K data points), relaxation delay 1 s, number of scans 8–16, number of increments 128–350 (depending on the spectral width to set the digital resolution around 6 Hz per point), zero filling to 2K data points and sine-bell resolution enhancement (sine-bell shift equal to half the acquisition time) in both dimensions.

Typical data for C,H-COSY (HETCOR direct detection, magnitude mode) were: acquisition time 0.05 s (2K data points), relaxation delay 1.5 s, number of increments 512, number of scans 80–200, zero filling to 4K points in F_2 and 2K points in F_1 and sine-bell resolution enhancement (sine-bell shift equal to half the aquisition time) in both dimensions

All pulse sequences used were delivered by the Varian standard software.

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